Noncrystalline Contents in Linear Polyethylene Samples, Crystallized from the Melt and Dilute Solution as Revealed by Proton Dipole Decoupled/Magic Angle Sample Spinning ¹³C NMR

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Summary

It is certified from identification of a chemical shift coinciding with O_{1SO} , a motionally averaged chemical shift observed for this polymer in solution that noncrystalline contents in both melt- and solution-grown crystals are in a similar chain conformation to that in the melt or solution. Furthermore, examination of magnetic relaxation times of the O_{1SO} line reveals that the local chain dynamics of the noncrystalline content are the same for both samples while a long range chain motion for the solution-crystal is somewhat restricted.

Introduction

Since lozenge-shaped crystallites were observed for solutiongrown polyethylene(Till 1957, Keller 1957, Fischer 1957), extensive studies have been carried out of the conformation and mode of molecular chains to participate in such crystallites with different techniques.* However, the final conclusion has been suspended because of the lack of a technique which could detect the chain conformation in noncrystalline materials in an order of molecule. Rather recently, however, new approaches to this problem have appeared with use of neutron scattering that principally can detect the chain conformation(Schelten et al. 1974). In relation to the chain conformation in solution-grown polyethylene, for instance, the mode of a molecular chain to participate in a given crystallite has been evaluated (Stamm et al. 1979, Yoon and Flory 1979, Sadler and Keller 1977). It is also concluded that the mean square radius of gyration of molecular chains in bulk-crystals is the same as in the melt (Stamm et al. 1979). These investigations have again stimulated researches of this problem with different techniques.

On the other hand, solid-state high-resolution ¹³C nmr that has recently been developed can detect the chemical shift of methylene carbon in solid polyethylene as a tensor quantity that is definable for each local chain conformation. Therefore, information of molecular chains in an order of molecule is expected by this technique, together with information of chain dynamics through examination of magnetic relaxation

^{*} See, for instance, Faraday Discussion of the Chemical Society, No.68 "Organization of Macromolecules in the Condensed Phase", Roy. Soc. Chem., London, 1979

times. This paper reports preliminary results of melt- and solution-grown polyethylene samples.

Experimentals

<u>Samples</u> Fractionated and unfractionated samples of linear polyethylene were used. <u>Bulk-Crystals</u>: Molecular weight fractions with M_V of 31,800 and 248,000 were isothermally crystallized at 130°C for 2-4 weeks from the melt after melting at 160°C for 5h under vacuum. An unfractionated sample with M_V of 3x10⁶ was isothermally crystallized at 129°C for 23 days after melting at 200°C for 1h under vacuum. <u>Solution-Crystal</u>: A molecular weight fraction with M_V of 106,000 was isothermally crystallized at 85°C for 24h from 0.08% tolulene solution.

 1^{3} C nmr Solid state 1^{3} C nmr was performed with JEOL JNM FX-100 under a static field of 2.37T;Larmor frequencies of 1^{3} C and 1H are 25.05 and 99.61 MH_z, respectively. Field strengths of H₁ were 6.0 and 1.5 mT for 1^{3} C and ¹H, respectively. The magic angle sample spinning was conducted throughout this work by the use of an Andrew type rotor of poly(chlorotrifluoroethylene) at a spinning rate of 3.2-3.5 kH_z. The pulse sequences utilized are described in appropriate places of the text. ¹H decoupling of a field strength of 1.5 mT was only added for data-sampling in all sequences. The chemical shift was calibrated with use of CH₂ line of adamantane with a chemical shift of 38.7 ppm from tetramethylsilane.

Results and Discussion

Indentification of Q_{ave} and Q_{iso} Fig. 1 shows the DD/MASS ¹³C nmr spectra of the bulk- and solution-crystals obtained by pulse sequences of $(\pi/4-1000s)_{64}$ and $(\pi/4-750s)_{84}$, respectively. These spectra are thought to reproduce completely contributions from all components in the structure.* The spectrum of the bulk-crystal evidently consists of two separable lines at chemical shifts of 34 and 32 ppm; the former coincides with O_{ave} (=1/3 Tr($\mathbf{0}$)), the average of diagonal elements of the chemical shift tensor for trans-trans chain conformation (Fyfe et al. 1978, Earl and VanderHart 1979, Schröter and Posern 1981), and the latter coincides with O_{iso} , a motionally averaged value observed for this polymer in solution. Hence, it is evident that the bulk-crystal contains a noncrystalline content, the chain conformation of which is similar to that in the melt or solution. On the other hand, the existence of σ_{iso} in the spectrum of the solution-crystal is somewhat vague, though it may be seen as a shoulder at 32 ppm. Since σ_{iso} represents noncrystalline contents in partially crystalline polyethylene, T1's at room temperature are

* Since T_1 of the rigid crystalline component of this polymer is to be about 200-300s(Schröter and Posern 1981), the waiting times longer than 750s (at least 2.5 times T_1) are enough for the equilibrium magnetization to recover more than 98% after each $\pi/4$ pulse. Furthermore, since the crosspolarization which may predominantly enhance the ^{13}C magnetization of rigid components is not used here, contributions from the mobile noncrystalline component are thought to be fully reproduced.



Fig. 1 DD/MASS ¹³C nmr spectra of polyethylene at room temperature. (a) the solution-crystal with \overline{M}_V of 106,000. (b) the bulk-crystal with \overline{M}_V of 3x10⁶. Pulse sequences are described in each spectrum. Dotted lines indicate constituent lines decomposed with a curve resolver.



Fig. 2 Partially relaxed DD/MASS ¹³C nmr spectra obtained by $(\pi - \tau - \pi/2)_X$ sequences with different τ 's, together with ¹H broad-line spectra of the bulk- and solution-crystals. Samples are the same as for Fig.1.

expected to be 0.2-0.3s from a scalar decoupled ¹³C nmr where only the rubbery component is detected (Kitamaru 1982). Furthermore, the T₁ of the rigid crystalline component which contributes to \mathcal{O}_{ave} is to be 200-300s (Schröter and Posern 1981). Accordingly, the existence of a noncrystalline content that contributes to \mathcal{O}_{iso} can be judged by examining partially relaxed spectra obtained by a sequence with a shorter waiting time $(\pi - \tau - \pi/2 - 10s)_x$ where only contributions of components with T₁'s shorter than 10s appear. Fig. 2 shows such partially relaxed spectra with different τ 's together with ¹H broad-line spectra of both samples. Here, the existence of the \mathcal{O}_{iso} line and faster recovery of the magnetization are evidently confirmed for both samples. See, for instance, the spectra of the solution-crystal at τ =1.00 and 2.00s. At τ =1.00s the magnetization of \mathcal{O}_{iso} has already recovered in the positive direction while that of \mathcal{O}_{ave} is still in the negative direction. At τ =2.00s only \mathcal{O}_{iso} is detected in the positive direction.

Based on the existence of both \mathcal{O}_{ave} and \mathcal{O}_{iso} lines, each spectrum in Fig. 1 was decomposed into two Lorentzians as shown by dotted lines in each spectrum. The integrated fraction of the \mathcal{O}_{ave} line is in accordance with the crystalline fraction of each sample estimated from density measurements as shown in Table 1. Hence, it is evident that the \mathcal{O}_{ave} and \mathcal{O}_{iso} are contributed from the crystalline and the entire noncrystalline components. This suggests that the entire noncrystalline component of both samples is in a similar local chain conformation to that in the melt or solution.

 $\frac{13C}{and} \frac{13C}{1981}$, the \mathcal{O}_{ave} line comprises two components with T_1 's of about 1-2s and 300s whereas the \mathcal{O}_{1SO} line comprises one component with a much shorter T_1 . Since it is difficult at present to measure exactly such a longer T_1 as 300s, we have estimated only T_1 's shorter than at least 10s with a pulse

TABLE I

Spin-lattice relaxation times of linear polyethlene samples obtained by DD/MASS ^{13}C NMR($\pi-\tau-\pi/2-10\text{s})_n$ and ^{1}H scalar-decoupled ^{13}C NMR at room temperature

Samples	crystalline fraction		T ₁ /s	
	$(1-\lambda)$ ¹	$(1-\lambda)0^{2}$	$\sigma_{ave}^{3)}$	Ø _{iso}
Bulk-Crystal fractionated My= 31,800	0.922		1.0	0.23
$M_V=248,000$ unfractionated $M_V=3\times10^6$	0.620	0.61	1.6	$(0.20)^{4}$ $(0.23)^{4}$
Solution-Crystal M _V =106,000	0.825	0.78	2.0	0.23

1.Crystalline fraction from density measurements. 2. Integrated fraction of \mathcal{O}_{ave} line in the total spectrum. 3.Among T₁'s of \mathcal{O}_{ave} , T₁'s shorter than 10s are detected. 4.T₁ estimated from scalar-decoupled ¹³C nmr.

sequence $(\pi-\tau-\pi/2-10s)_{\mathbf{X}}$. The data of samples with different molecular weights and modes of crystallization are filed in Table 1. As can be seen, the T_1 's of O_{iso} are virtually equivalent for all samples regardless of molecular weights and modes of crystallization.* These T1's are in good accordance with T1's obtained by a scalar-decoupled ¹³C nmr for this polymer at room temperature. This coincidence is meaningful because it suggests that all noncrystalline contents can be detected also by the scalar-decoupled ¹³C nmr at least at room temperature. In any case, it is concluded that the entire noncrystalline content in crystalline polyethylene samples is not only in a chain conformation similar to that in the melt but also it is associated with an equivalent spin-lattice relaxation time, whether they are crystallized from the melt or dilute solution. However, this does not necessarily mean that the chain dynamics of the noncrystalline content are the same independent of the morphology. See, for instance, the ¹H broad-line spectra of the bulk- and the solution-crystal shown in Fig. 2. The linewidth of the noncrystalline component of the solution-crystal is much wider than that of the bulk-crystal.** It is also confirmed that the linewidth of scalar-decoupled ¹³C nmr spectrum of the solution-crystal is much wider than that of the bulkcrystal. These results suggest that the noncrystalline content is associated with an equivalent T_1 but different T_2 's depending on the morphology of samples; the T_2 of the solutioncrystal is much shorter than that of the bulk-crystal.

In regard to shorter T_1 's of \mathcal{O}_{ave} in an order of 1-3s, they will stem from some mobility within the crystalline region, because the integrated fraction of the \mathcal{O}_{ave} line coincides with the crystalline fraction estimated from density measurements for both samples. It should be assumed that such a chain motion as 180° -jump rotation of crystalline sequences somewhat occurs at room temperature, by the anology of the case of nhydrocarbons prior to melting(Strobl 1977).

<u>Chain dynamics of noncrystalline contents</u> The noncrystalline contents of the bulk- and solution-crystals are associated with an equivalent T_1 and different T_2 's. This apparently contradicts the suggestion from the single-correlation-time theory (Kubo and Tomita 1954, Solomon 1955) where the internuclear vector connecting coupled ¹³C and ¹H is assumed to fluctuate spherically at random with one correlation time. So far as we persist in this theory, this contradiction cannot be solved even if a wide distribution in the correlation time is introduced. However, as will be discussed in detail elsewhere, if we assume plural correlation times at least more than three, this contradiction is easily solved and important information of the chain motion is obtainable (Kitamaru et al. 1982).

The equivalent T_1 of \mathcal{O}_{iso} for both bulk- and solutioncrystals indicates a similar local chain motion of the noncrystalline contents in addition to a similar local chain con-

^{*} Such an independence of T_1 's on morphology of samples is also confirmed by Komoroski and Mandelkern(1977) for T_1 's obtained by a scalar-decoupled ¹³C nmr.

^{**} For detailed analysis of the broad-line nmr spectrum of this polymer, see the article by Kitamaru and Horii(1978).

formation to those in the melt. Different T_2 's indicate a difference in a long range chain motion between two samples; the long range motion of noncrystalline molecular chains in the solution-grown sample is restricted because the both ends of chains are terminated by crystalline sequences in a given crystallite.

It is proposed from neutron scattering analysis of solution-grown samples that a molecular chain participates repeatedly in a given crystallite, forming "a stacked sheet" in the direction parallel to the (110) crystal plane(Stamm et al. 1979). Regardless of validity of this stacked model, if a molecular chain is assumed to participate in a given crystallite with a thickness of about 100Å, noncrystalline sequences in samples with a crystalline fraction of 0.8 must consist of methylene sequences of about 20 CH₂ repeating units, both ends of which are terminated with crystalline sequences. How are the conformation and dynamics of such a short noncrystalline sequence? It is significant that such a short methylene sequence can involve rather variable chain conformation as has been suggested by this nmr work.

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